

Preliminary communication

DERIVATIVES OF THE *closo*-Ru₄C₂ CLUSTER

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Summary

The reaction of Ru₃(CO)₁₂ with RC≡CR' yields *closo*-Ru₄(CO)₁₂ (RC≡CR'), which reacts with additional alkyne, PhC≡CR'', to yield Ru₄(CO)₁₁(RC≡CR')(PhC≡CR'') (PhC≡CR'').

In previous communications [1,2] we reported the production of organometallic compounds based on *closo*-Ru₄C₂ units from the reaction of H₄Ru₄(CO)₁₂ and cyclic olefins. Few compounds of this class are known; the only other example, apart from those of ruthenium, being Co₄(CO)₁₀C₂R₂ [3], and in the case of Ru₄(CO)₁₁C₈H₁₀ [4] it has been shown that the Ru₄C₂ skeleton is essentially the same as that of Co₄C₂ found in Co₄(CO)₁₀C₂Et₂ [3]. This paper reports the preparation of several additional compounds based on *closo*-Ru₄C₂ units which are formed in the reaction of Ru₃(CO)₁₂ with the alkynes RC≡CR' (R = Ph, R' = Ph, Me or Et; R = R' = CH₃OCH₂).

The cluster compounds Ia–Id are readily prepared by direct reaction of the appropriate alkyne with Ru₃(CO)₁₂ in refluxing n-hexane under argon, usually in yields in the region of 5–10%. (Various other, as yet uncharacter-

SCHEME 1. THE REACTION OF Ru₃(CO)₁₂ WITH THE ALKYNES RC≡CR'

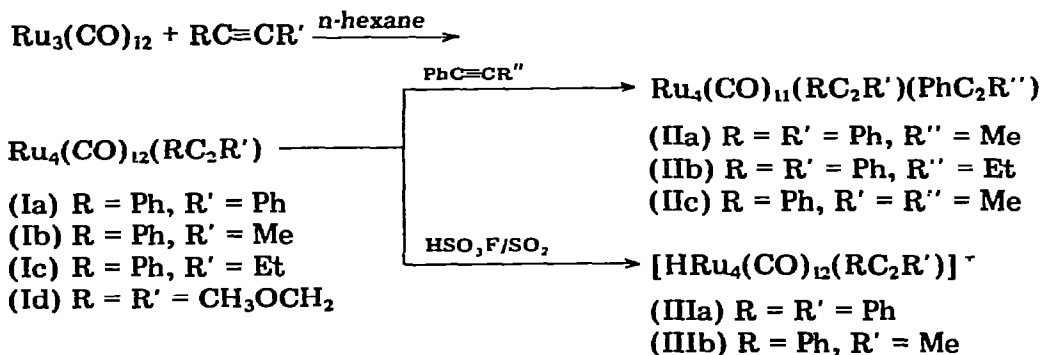


TABLE 1
INFRARED AND MASS SPECTROSCOPIC DATA FOR *closo*-Ru₄C₂ COMPOUNDS

Compound	IR (in n-pentane) (cm ⁻¹)	M ⁺ (m/e)
Ia	2091m, 2065vs, 2041(sh), 2038vs, 2017m, 2010(sh), 1969m	922
Ib	2091m, 2064vs, 2039(sh), 2034vs, 2013m, 2003(sh), 1968m	860
Ic	2090m, 2062vs, 2040(sh), 2037vs, 2012m, 2003(sh), 1971m	874
Id	2091m, 2063vs, 2037vs, 2012m, 2006(sh), 2003(sh), 1970m	858
Ru ₄ (CO) ₁₂ C ₈ H ₁₀ [2]	2090m, 2061vs, 2038s, 2035s, 2013w (br), 2003m	
Ru ₄ (CO) ₁₂ C ₈ H ₁₂ [2]	2088w, 2059s, 2034s, 2010m, 1996m, 1967w (br)	
IIa	2083vs, 2053vs, 2035s, 2024s, 2017(sh), 1978vs	1010
IIb	2083vs, 2054vs, 2038(sh), 2034s, 2024s, 2015(sh), 1978s, 1974(sh)	1024
IIc	2081vs, 2052vs, 2034s, 2024s, 2011s, 1977s	948
Ru ₄ (CO) ₁₁ C ₈ H ₁₀ [2]	2081m, 2046vs, 2029s, 2019m, 2007w, 1999w, 1990w (br), 1966w (br)	

ised, clusters are also obtained.) Separation of these products is relatively easy and is best carried out by thin layer chromatography on silica. Scheme 1 lists the new compounds prepared and some spectroscopic data are collected in Table 1. All complexes are intensely coloured (brown) as observed with other Ru₄C₂ derivatives [1,2] and not yellow as generally observed for clusters based on the Ru₃ unit [1,2].

The formulation of the compounds as derivatives of *closo*-Ru₄C₂ (Fig.1) is based on mass spectroscopic evidence, supported by the appropriate analytical and infrared spectroscopic data. Thus all complexes exhibit a strong parent ion in their mass spectra (see Table 1) together with ions corresponding to the stepwise loss of the twelve carbonyl groups and fragmentation of the RC≡CR' ligand. An intense peak corresponding to the Ru₄C₂⁺ ion is also observed; an observation consistent with the presence of a *closo*-Ru₄C₂ unit within the complex. The infrared spectra (Table 1) in the metal carbonyl region ($\nu(\text{CO})$) of complexes Ia-Id are very similar to those of Ru₄(CO)₁₂C₈H₁₀ and Ru₄(CO)₁₂C₈H₁₂ [2]. They do not resemble those of the trinuclear iron compounds Fe₃(CO)₉(RC≡CR') [4]. From their formulae and spectroscopic properties these compounds are obviously similar to those formed from H₄Ru₄(CO)₁₂ and cyclic olefins. The formation of Ru₄ clusters from Ru₃(CO)₁₂ is perhaps unexpected, particularly in view of previous work with Fe₃(CO)₁₂ [5] and Os₃(CO)₁₂ [6] which showed that on reaction with acetylenes these carbonyls gave predominantly trinuclear species. In a recent communication Cetini and his coworkers [7] reported the formation of Ru₃ trimers and dimers

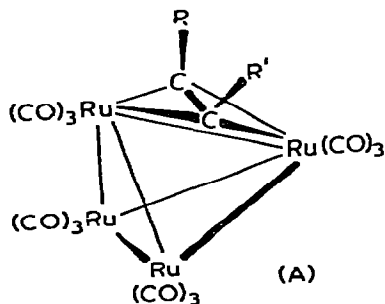


Fig.1. Probable structure of Ru₄(CO)₁₂(RC₂R').

from the reaction of $\text{Ru}_3(\text{CO})_{12}$ and diphenylacetylene or tetraphenylcyclopentadienone under conditions similar to those reported here. No mention of tetranuclear Ru_4 derivatives was made.

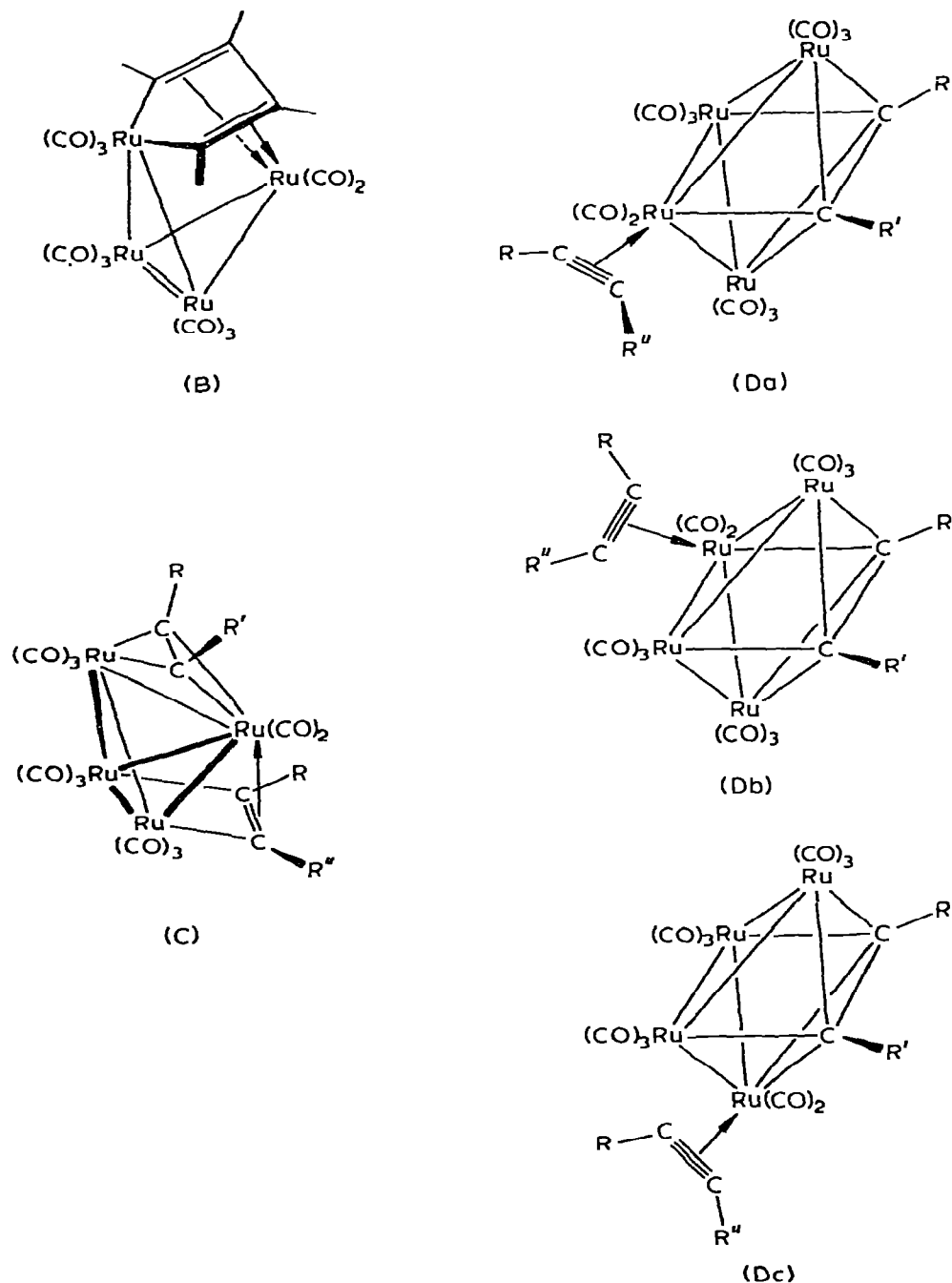


Fig. 2. Possible structures of $\text{Ru}_4(\text{CO})_{11}(\text{RC}_2\text{R}')(\text{PhC}_2\text{R}'')$.

On allowing complexes Ia or Ib to react with additional alkyne ($\text{PhC}\equiv\text{CR}''$ in n-hexane) ($\text{R}'' = \text{Me}$ or Et), (see Scheme 1) under argon (reflux 6 h), carbon monoxide is evolved and new complexes based on a Ru_4 cluster unit are produced, which are obtained in yields of 50-60%, after TLC on silica. These new orange complexes are soluble in a wide range of organic solvents. On the basis of their mass spectra, which exhibit strong parent ions, they may be formulated as $\text{Ru}_4(\text{CO})_{11}(\text{PhC}\equiv\text{CR}')(\text{PhC}\equiv\text{CR}'')$. This formulation is supported by analytical data and their infrared spectra which are similar to those reported for other $\text{Ru}_4(\text{CO})_{11}$ species [2]. Their structures are unknown. Several possibilities are given in Fig.2. At present we have no way of differentiating these possibilities apart from the observation that a peak corresponding to Ru_4C_4^+ appears in the mass spectrum, which tends to imply that the two acetylenes have combined [8] thereby favouring structural possibility B.

TABLE 2

¹H NMR SPECTRA OF COMPLEXES (IIIa) AND (IIIb) IN LIQ. $\text{SO}_2 + 1$ drop HSO_3F^a

Compound	Temp. ($^\circ\text{C}$) ^b	Chemical shifts (τ , ppm)
IIIa	+10	2-3m (10H), 33.4s (1H)
IIIa	-10	2-3m (10H), 33.6s (1H)
IIIa	-40	2-3m (10H), 33.8s (1H)
IIIa	-60	2-3m (10H), 33.9s (1H)
IIIb	0	2-3m (5H), 6.3s (3H), 32.8s (1H)
IIIb	-20	2-3m (5H), 6.3s (3H), 32.8s (1H)
IIIb	-60	2-3m (5H), 6.3s (3H), 32.8s (1H)

^a m = multiplet, s = singlet; relative intensities in parentheses. ^b ± 0.2 $^\circ\text{C}$.

Complexes Ia and Ib also undergo protonation in strong acid (see Scheme 1) to yield the hydrido cations $[\text{HRu}_4(\text{CO})_{12}(\text{RC}_2\text{R}')]^+$. The hydrido ligand was detected from the ¹H NMR spectrum (Table 2) and clearly must span a Ru-Ru edge in the cluster, although facile migration of H ligands from one polyhedral edge to another is commonly observed [9-11] for organometallic complexes of this type.

Hydrogenation of $\text{Ru}_4(\text{CO})_{12}\text{PhC}_2\text{Ph}$ in n-heptane under reflux (15 min) gives an almost quantitative yield of $\text{H}_4\text{Ru}_4(\text{CO})_{12}$ and *trans*-stilbene. The mechanism of this hydrogenation will be discussed in a subsequent paper.

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